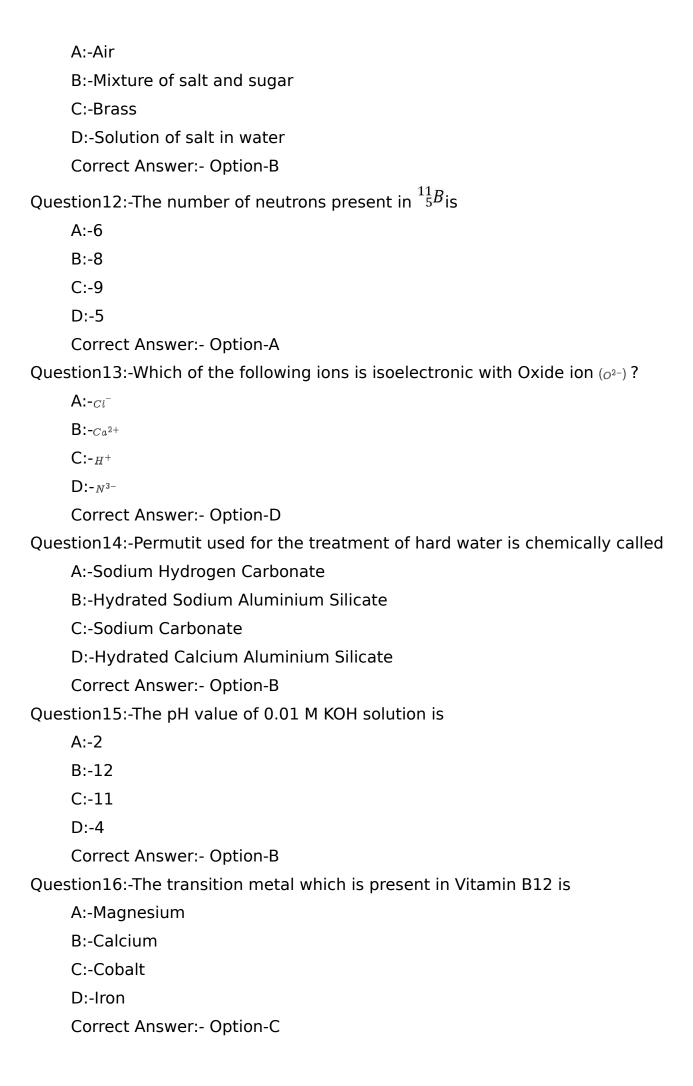
FINAL ANSWER KEY

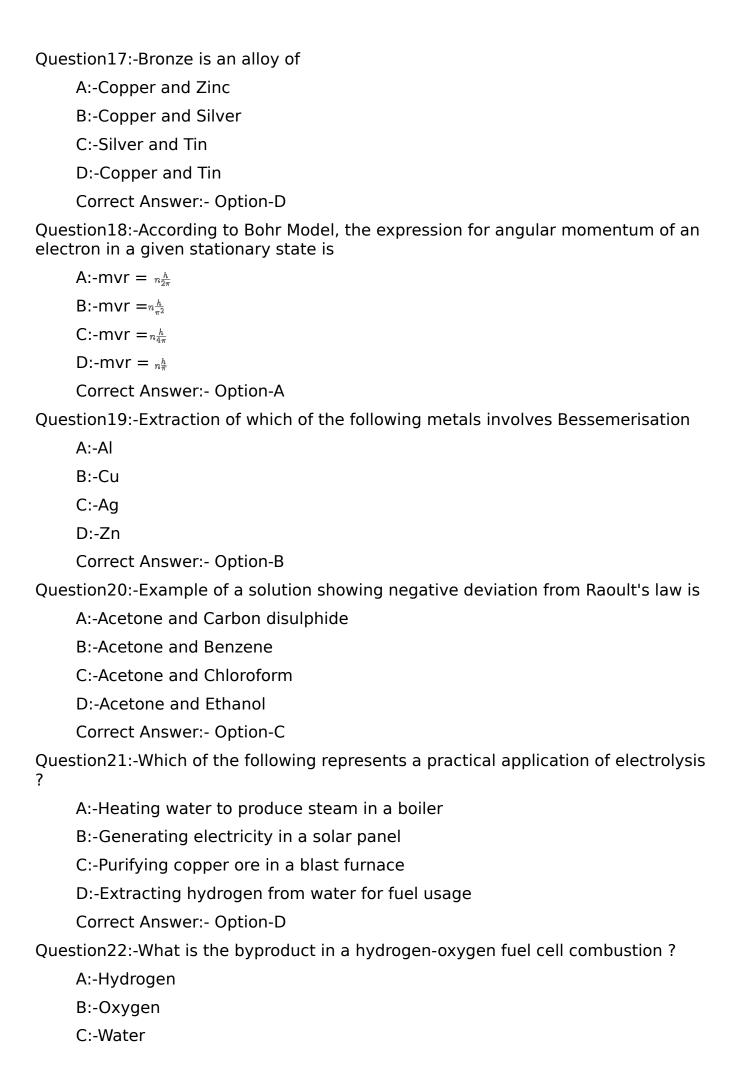
Question 114/2024/OL Paper Code: Category 669/2023 Code: Exam: Junior Instructor (Electroplater) Date of Test 10-10-2024 Department Industrial Training Question1:-Which electrical quantity cannot be measured with a voltmeter? A:-voltage B:-potential difference C:-terminal voltage D:-EMF Correct Answer:- Option-D Question2:-Which effect of electrical current is applied in fan? A:-heating B:-magnetic C:-chemical D:-gas ionisation Correct Answer:- Option-B Question3:-Which electrical quantity will be the same at any point of the series circuit? A:-voltage B:-current C:-resistance D:-power Correct Answer:- Option-B Question4:-Which one is the source of ac supply? A:-cell B:-battery C:-DC generator D:-alternator Correct Answer:- Option-D Question5:-What is the value of conductance, if a wire has resistance of R Ω ? A:-R

B:-1/R

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D:-R/2
    Correct Answer:- Option-B
Question6:-What is the Quantity of electricity transferred by one ampere current in
one second?
     A:-one volt
    B:-one watt
     C:-one coulomb
    D:-one joule
     Correct Answer:- Option-C
Question7:-Which material permit electrones to move through it easily?
    A:-eureka
     B:-silicon
    C:-air
    D:-mica
     Correct Answer:- Option-A
Question8:-What is the effect of the parallel circuit with one branch open?
     A:-Whole circuit will not function
     B:-Voltage increases in that branch
     C:-Current will be the same
    D:-No current will flow in that branch
     Correct Answer: - Option-D
Question9:-How many electrons are there in the valance shell of the aluminium
atom?
     A:-1
    B:-4
    C:-3
    D:-2
     Correct Answer:- Option-C
Question10:-Ohm's law shows the relation between
    A:-current, voltage, power
     B:-current, voltage, resistance
     C:-conductance, voltage, resistance
    D:-current, emf, voltage
     Correct Answer:- Option-B
Question11:-Which of the following is not a homogenous mixture?
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C:-2R





D:-None of the above

Correct Answer:- Option-C

Question23:-What happens when the opposing external applied potential to an electrochemical cell is greater than the cell potential

A:-The electrochemical cell stops functioning

B:-The electrochemical cell behaves like an electrolytic cell

C:-Only reduction reactions occur in the cell

D:-Only oxidation reactions occur in the cell

Correct Answer:- Option-B

Question24:-Calculate the quantity of aluminium metal deposited by passing 1.5 faradays of electricity through an aqueous solution of $AlCl_3$

A:-13.5g

B:-27g

C:-40.5g

D:-1.35g

Correct Answer:- Option-A

Question25:-The electrolyte used in Leclanche cell

A:-A paste of KOH and ZnO

B:-38% solution of sulphuric acid

C:-Moist sodium hydroxide

D:-Moist paste of NH_4Cl and $ZnCl_2$

Correct Answer:- Option-D

Question26:-Pick out the weak electrolytes from the following

- i) CH3COONa
- ii) NH4OH
- iii) NH_4Cl
- iv) CH3COOH

A:-i and ii

B:-i and iii

C:-ii and iii

D:-ii and iv

Correct Answer: - Option-D

Question27:-The product formed at the cathode during the electrolysis of molten sodium chloride

A:-Chlorine gas

B:-Sodium metal

C:-Hydrogen gas

D:-Oxygen gas

Question28:-In a parallel arrangement, two batteries with EMFs of 2V and 1V and internal resistance of 1Ω and 2Ω respectively, are connected. What is the effective EMF of the combination ?

A:-3/5V

B:-3/2V

C:-5/3V

D:-2V

Correct Answer:- Option-C

Question29:-Calculate the amount of charge flowing in 2 min in a wire of resistance 10Ω when a potential difference of 20V is applied between its ends

A:-120C

B:-240C

C:-20C

D:-4C

Correct Answer:- Option-B

Question30:-Energy change occurring in an electrolytic cell is

A:-Chemical energy to electrical energy

B:-Electrical energy to chemical energy

C:-Chemical energy to heat energy

D:-None of the above

Correct Answer:- Option-B

Question31:-Factors which affecting the atmospheric corrosion

- i) Oxidising action of air.
- ii) Humidity condition and atmospheric pollution.
- iii) Salt in marine atmosphere.

A:-Only (ii and iii)

B:-Only (i and ii)

C:-All of the above (i, ii and iii)

D:-Only (iii)

Correct Answer:- Option-C

Question32:-The corrosion of metals can be effectively resisted by

- i) Application of metallic coatings.
- ii) Application of acids.
- iii) Application of paints.

A:-Only (i and iii)

B:-Only (i and ii)

C:-Only (ii and iii)

D:-All of the above (i, ii and iii)

Question33:-A discontinuity in protective surface film

A:-Inter granular corrosion

B:-Pitting corrosion

C:-Stress corrosion

D:-Bacterial corrosion

Correct Answer:- Option-B

Question34:-Cracking of light alloys

A:-Inter granular corrosion

B:-Pitting corrosion

C:-Stress corrosion

D:-Bacterial corrosion

Correct Answer:- Option-C

Question35:-The aluminium surface can be protected from corrosion by

A:-Pickling

B:-Colouring

C:-Dyeing

D:-Anodising

Correct Answer:- Option-D

Question36:-The corrosion caused by moving liquids

A:-Filiform corrosion

B:-Fretting corrosion

C:-Corrosion fatigue

D:-Erosion corrosion

Correct Answer:- Option-D

Question37:-The coating of zinc metal on iron surface for the protection from corrosion

A:-Parkerising

B:-Galvanising

C:-Dyeing

D:-Anodising

Correct Answer:- Option-B

Question38:-In which corrosion a growth of crack formation in a corrosive environment on ferrous metal?

A:-Inter granular corrosion

B:-Pitting corrosion

- C:-Stress corrosion
- D:-Bacterial corrosion

Question39:-Which of the following chemical used for the removal of corrosion in aluminium metal?

- A:-Sodium hydroxide
- B:-Hydrochloric acid
- C:-Nitric acid
- D:-Aquaregia

Correct Answer:- Option-A

Question 40:- Why aluminium oxide on the surface of aluminium useful for the aluminium metal?

- A:-It helps the oxidation process of aluminium
- B:-It helps the formation of new aluminum metal
- C:-It protects the aluminium oxide film
- D:-It protects further deterioration of the aluminium metal

Correct Answer: - Option-D

Question41:-Mild steel immersion heaters are used for heating electroplating solutions

- i) Cyanide solutions
- ii)Alkaline solutions
- iii) Phosphoric and chromic acid solutions
 - A:-Only (i and iii)
 - B:-Only (i and ii)
 - C:-Only (ii and iii)
 - D:-All of the above (i, ii and iii)

Correct Answer:- Option-B

Question42:-Chrofles are used to float the plating solution for

- i) Improve throwing power
- ii) Reduce spray
- iii) Prevent heat losses
 - A:-Only (i and iii)
 - B:-Only (i and ii)
 - C:-Only (ii and iii)
 - D:-All of the above (i, ii and iii)

Correct Answer:- Option-C

Question43:-In plating solution cooling purpose is required

- i) Operated at high current densities
- ii) Operated at high temperature
- iii) Operated at low temperature

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A:-Only (i)
     B:-Only (i and ii)
     C:-Only (ii and iii)
     D:-Only (iii)
     Correct Answer:- Option-B
Question44:-What is the reason for the proposal of air agitation units in plating bath
i) Keep the solution well mixed
ii) Prevent the production of rough or burnt deposit
iii) Increase the metal content
     A:-Only (i and iii)
     B:-Only (i and ii)
     C:-Only (ii and iii)
     D:-Only (iii)
     Correct Answer:- Option-B
Question45:-Where it is recommended to use filter units to plating bath?
i) Roughness of deposit due to inclusion of suspended matters
ii) Roughness of deposit due to high metallic content
iii) Roughness of deposit due to high current density
     A:-Only (i)
     B:-Only (i and ii)
     C:-Only (ii and iii)
     D:-Only (iii)
     Correct Answer: - Option-A
Question 46: The effective method for the rapid drying of small plated components
in bulk?
i) By the use of hot air oven
ii) By the use of hot water
iii) By the use of centrifugal dryer
     A:-Only (i)
     B:-Only (i and ii)
     C:-Only (ii and iii)
     D:-Only (iii)
     Correct Answer:- Option-D
Question47:-The advantages of horizontal immersed perforated plating barrels
i) Easily provided with automatic loading and unloading facilities.
ii) Easily taken from pre-plating, plating and post plating treatments.
iii) Convenient for the plating of large quantity in bulk.
     A:-Only (i)
     B:-All of the above (i, ii and iii)
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C:-Only (ii and iii)

D:-Only (iii)

Correct Answer:- Option-B

Question 48: The equipment in which the abrasive particles are made to impinge on the surface to be cleaned by compressed air?

A:-Buffing

B:-Emery cutting

C:-Barrel cleaning

D:-Abrasive blast cleaning

Correct Answer:- Option-D

Question49:-Name of the cathode contact used in plating barrels consists of a length of heavy insulated flexible cable terminated with a steel cathode knob.

A:-Dummy

B:-Bus bar

C:-Danglers

D:-Jigs

Correct Answer:- Option-C

Question50:-Name of the instrument used to measure accurately the acidic or alkalinity of a solution

A:-pH meter

B:-Hydrometer

C:-Lactometer

D:-Thermometer

Correct Answer:- Option-A

Question51:-Which solution is used for pickling of non-ferrous metals?

A:-Sulphuric acid solution

B:-Hydro fluoric acid solution

C:-Nitric acid solution

D:-Hydro chloric acid solution

Correct Answer:- Option-B

Question52:-What is a buffing process?

A:-Coating

B:-Free-cleaning

C:-Mechanical cleaning

D:-Ionising

Correct Answer:- Option-C

Question 53:- Which is the process included in the pre-treatment process?

- A:-Backing
- B:-Degreasing and Pickling
- C:-Polishing and buffing
- D:-Passivation

Question54:-What is the removal of maskant on the areas to be etched?

- A:-Demasking
- B:-Etching
- C:-Masking
- D:-Scribing

Correct Answer:- Option-A

Question55:-What is the mixture of concentrated nitric acid and same concentration of hydrochloric acid and is used for dissolving metals?

- A:-Oil of vitriol
- B:-Schiff's reagent
- C:-Aqua regia
- D:-Grignard reagent

Correct Answer:- Option-C

Question 56:-Which component is essential in an automatic barrel plating plant for holding the parts ?

- A:-Rack
- B:-Basket
- C:-Barrel
- D:-Conveyor

Correct Answer:- Option-C

Question57:-Which method provides quick and complete cleaning for parts to be plated or coated, saving time and ensure proper adhesion ?

- A:-Soak cleaning
- B:-Buffing
- C:-Barel Polishing
- D:-Ultrasonic cleaning

Correct Answer:- Option-D

Question58:-Which one of the following is an organic chemical?

- A:-Sodium Hydroxide
- B:-Tri chloroethylene
- C:-Zinc Sulphate
- D:-Pottassium Permanganate

Question59:-What is the first step in a chemical cleaning cycle employed for heavily contaminated items?

A:-Solvent cleaning

B:-Vapour cleaning

C:-Ultrasonic cleaning

D:-Electrolytic cleaning

Correct Answer:- Option-A

Question60:-Which is the process in which the abrasive particles are made to impinge on the surface to be cleaned by compressed air?

A:-Buffing

B:-Emery cutting

C:-Abrasive blast cleaning

D:-Barrel cleaning

Correct Answer:- Option-C

Question61:-Which chemical is used for wood's nickel strike process?

A:-Nickel Chloride and Hydro chloric acid

B:-Nickel sulphate and Hydro chloric acid

C:-Nickel chloride and Sulphuric acid

D:-Nickel sulphate and Sulphuric acid

Correct Answer:- Option-A

Question62:-Which anodes are used for bright chromium plating?

A:-Chromium anode

B:-Antimony lead anode

C:-Stainless steel anode

D:-Nickel anode

Correct Answer:- Option-B

Question63:-Which under coating is to be done before gold plating?

A:-Chromium

B:-Nickel

C:-Cadmium

D:-Zinc

Correct Answer:- Option-B

Question64:-Which process removes Zinc cadmium and copper contamination on the nickel bath?

A:-Filtration

B:-Activated carbons

- C:-Plating off
- D:-Agitation

Question65:-Which type of plating involves the use of electrolytes containing copper cyanide and zinc cyanide?

- A:-Nickel plating
- B:-Zinc plating
- C:-Copper plating
- D:-Brass plating

Correct Answer:- Option-D

Question66:-Which is the Latin word from which the symbol of Tin (Sn) is derived?

- A:-Aurum
- B:-Stannum
- C:-Argentum
- D:-Cuprum

Correct Answer:- Option-B

Question67:-What are the chemical contains in electrolyte used for bright nickel plating?

- A:-Nickel chloride, Boric acid and Ammonium sulphate
- B:-Nickel sulphate, Nickel chloride, Boric acid
- C:-Dilute sulphuric acid, Hydro chloric acid and Nickel chloride
- D:-Nickel sulphate, Sulphuric acid and Ammonium sulphate

Correct Answer:- Option-B

Question68:-Which is used as the main electrolytic ingredient in Trivalent chromium plating?

- A:-Chromium Sulphate
- B:-Sulphuric acid
- C:-Chromium Oxide
- D:-Phosphoric acid

Correct Answer:- Option-A

Question69:-Which is the cathode efficiency of alkaline Tin plating solution?

- A:-98 to 100%
- B:-70 to 80%
- C:-45 to 60%
- D:-90 to 95%

Correct Answer:-Question Cancelled

Question 70:- Which plating can be used in food packages?

- A:-Zinc plating
- B:-Tin plating
- C:-Cadmium plating
- D:-Brass plating

Question71:-Which acid is silver not resistant to?

- A:-Sulphuric acid
- B:-Nitric acid
- C:-Citric acid
- D:-HCL acid

Correct Answer:- Option-B

Question72:-What is necessary to keep the plating solution well mixed inorder to prevent variation in concentration and temperature?

- A:-Agitation
- **B:-Degreasing**
- C:-Filtration
- D:-Pickling

Correct Answer: - Option-A

Question73:-Which chemical should be added in cyanide copper plating bath if the pH is too low?

- A:-Hydrochloric acid
- B:-Acetic acid
- C:-Sodium Hydroxide
- D:-Sodium cyanide

Correct Answer:- Option-C

Question74:-What is the possible cause for the deposited blistered or stripping from the basis metal while doing bright chromium plating?

- A:-Good adhesion of the undercoat nickel deposit
- B:-Poor adhesion of the undercoat nickel deposit
- C:-Poor adhesion of the undercoat chromium deposit
- D:-Good adhesion of the undercoat chromium deposit

Correct Answer:- Option-B

Question75:-What is the possible cause of hard and dull deposit in silver plating?

- A:-Too low current density
- B:-High current density
- C:-Too low temperature
- D:-High temperature

Question 76:-Which process is used for preventing discolouration after cadmium plating ?

A:-Activation

B:-Phosphating

C:-Passivation

D:-Coating

Correct Answer:- Option-C

Question77:-Which solution is used to test for porosity of cadimum and zinc coating ?

A:-3% volume of hydrochloride acid solution

B:-2% volume of hydrochloric acid solution

C:-4% volume of hydrochloric acid solution

D:-1% volume of hydrochloric acid solution

Correct Answer:- Option-D

Question78:-Which defect occurs due to inefficient contacts of articles in bright chromium plating?

A:-Dull, grey, rough deposit

B:-Brown stains on deposit

C:-Blistered deposits

D:-Little chromium deposit on the article

Correct Answer:- Option-B

Question79:-Which is used in plating solutions as complexing agent particularly in gold and silver plating ?

A:-Buffer

B:-Cyanide

C:-Source of metal

D:-Brightner

Correct Answer:- Option-B

Question80:-What one of the following offers excellent solderability, as well as superior protection against corrosion?

A:-Steel

B:-Zinc

C:-Copper

D:-Tin

Correct Answer:- Option-D

Question81:-What is the chemical formula of aluminum?

	A:-AI
	B:-AM
	C:-Ag
	D:-Au
	Correct Answer:- Option-A
Que	stion82:-Which process is used to prevent corrosion of Aluminum?
	A:-Stripping
	B:-Blackening
	C:-Degreasing
	D:-Anodising
	Correct Answer:- Option-D
	stion83:-The coating formed by anodising of aluminium by sulphuric acid tess mainly consists of
	A:-Hydroxide of Aluminium
	B:-Oxide of Aluminium
	C:-Sulphate of Aluminium
	D:-Carbonate of Aluminium
	Correct Answer:- Option-B
Que	stion84:-Anodising of the following metal is not possible?
	A:-Titanium
	B:-Aluminium
	C:-Magnesium
	D:-Iron
	Correct Answer:- Option-D
Que	stion85: is normally used for sealing of anodised Aluminium ?
	A:-Nickel Chloride
	B:-Hydrochloric Acid
	C:-Nickel Acetate
	D:-Sulphuric Acid
	Correct Answer:- Option-C
	stion86:-Which of the following solution is not suitable for anodising of ninium ?
	A:-Hydrochloric Acid
	B:-Sulphuric Acid
	C:-Chromic Acid
	D:-Sulphuric Acid and Oxalic Acid

Correct Answer:- Option-A Question87:-In anodising of Aluminium, dye is used ? A:-Before Anodising **B:-After Sealing** C:-In the anodising bath D:-After anodising Correct Answer:- Option-D Question88:-For stripping of anodised film of aluminium, the following solution is used ____ ? A:-Chromic acid and phosphoric acid B:-Potassium dichromate C:-Nitric acid and acetic acid D:-Nickel acetate Correct Answer:- Option-A Question89:-Which process is followed for closing the pores in anodised films? A:-Passivation B:-Buffing C:-Sealing D:-Tumbling Correct Answer:- Option-C Question 90:-What is the process of applying colour to an anodised surface called? A:-Coating B:-Dyeing C:-Electroplating D:-Printing Correct Answer:- Option-B Question91:-How will you increase the adhesion property of the base metal? A:-By Pretreatment B:-By Increasing the size of bath C:-By passing more current D:-By improving pH value of the electrolyte Correct Answer:- Option-A Question 92:-Which one of the following property is tested using BNF test? A:-Hardness B:-Thickness of Electroplating deposit C:-Flexibility

D:-Tempering
Correct Answer:- Option-B
Question93:-Salt spray test is used to determine the of the metalic surface ?
A:-Hardness
B:-Corrosion resistance property
C:-Emissivity
D:-Wear resistance property
Correct Answer:- Option-B
Question94:-Tape Test is carried to measure the of coating ?
A:-Thickness
B:-Purity
C:-Adhesion
D:-Emmissivity
Correct Answer:- Option-C
Question95:-Which type of measuring instrument from the following is used for measuring the thickness of deposit in the cylindrical job?
A:-Micrometer
B:-Steel Rule
C:-Weighing Machine
D:-Measuring tape
Correct Answer:- Option-A
Question96:-Which method is used to measure the thickness of anodised film of aluminum?
A:-Ultrasonic
B:-Eddy Current
C:-Magnetic
D:-None
Correct Answer:- Option-B
Question97:-The method used for determining the thickness of metallic coating by electro chemical stripping is known as?
A:-Corrodkote test
B:-Tape test
C:-Salt spray test
D:-Coulometric test
Correct Answer:- Option-D
Question98:-What is the use of salt spray test for plated article?
A:-Adhession

- B:-Thickness
- C:-Hardness
- D:-Porosity

Question99:-What is the common visual test used to determine cleanliness of the surface of parts?

- A:-Tape test
- B:-Breakdown voltage test
- C:-Water break test
- D:-Salt spray test

Correct Answer:- Option-C

Question100:-Which corrosion resistance test is used to simulate the type of corrosion occure on Nickel Chromium coating?

- A:-Calcium test
- B:-Corrodkote test
- C:-Salt spray test
- D:-Sulphur-di-oxide test

Correct Answer:- Option-B