FINAL ANSWER KEY

Question1:-Who among the following missionary opened an English school at Mattanchery in 1818 with the financial aid from the Cochin Government ?

A:-Rev. J. Dawson

B:-Rev. Mead

C:-W.T. Ringletaube

D:-Bailey

Correct Answer:- Option-A

Question2:-Who authored the book Mokshapradipam ?

A:-Brahmananda Sivayogi

B:-Sri Narayanaguru

C:-Chattampi Swamikal

D:-Swami Vagbhatananda

Correct Answer:- Option-A

Question3:-Guruvayur Satyagraha was started under the auspices of the

A:-Nair Service Society

B:-Yogakshema Sabha

C:-Kerala Provincial Congress Committee

D:-Communist Party of India

Correct Answer:- Option-C

Question4:-In which year Ayyankali founded the Sadhu Jana Paripalana Yogam ?

A:-1903 A.D.

B:-1907 A.D.

C:-1910 A.D.

D:-1917 A.D.

Correct Answer:- Option-B

Question5:-Name the Diwan who issued order permitting the converted Shanar women to cover their bodies with jackets ?

A:-Velu Thampi

B:-Ummini Thampi

C:-Padmanabhan Menon

D:-Col. Munro

Correct Answer:- Option-D

Question6:-Where did Royal Indian Navy (RIN) revolt start in February 1946 ?

A:-Calcutta

B:-Karachi

C:-Karwar

D:-Bombay

Correct Answer:- Option-D

Question7:-Which Article of Indian Constitution prohibits untouchability ?

A:-Article 16

B:-Article 17

C:-Article 18

D:-Article 19

Correct Answer:- Option-B

Question8:-The World Economic Forum 2018 was held at

A:-Paris

B:-New York

C:-Davos

D:-New Delhi

Correct Answer:- Option-C

Question9:-Who was the Second Vice Chairman of the Higher Education Council of Kerala ?

    A:-Dr. B. Iqbal

    B:-Dr. K.N. Panikkar

    C:-T.P. Sreenivasan

    D:-Dr. Rajan Gurukkal

    Correct Answer:- Option-C

Question10:-Where did India's first U.N. Technology innovation lab set up ?

    A:-Mumbai

    B:-Thiruvananthapuram

    C:-New Delhi

    D:-Bangalore

    Correct Answer:- Option-B

Question11:-The Inclusive Education approach must have

 (i) elements of social justice and equitable distribution

 (ii) more autonomy for teachers and students

(iii) care and development of varying ability among the student population

(iv) participation of physically and intellectually challenged students in normal setting

 (v) single examination for all children

    A:-(i) and (iv) only

    B:-(iii) and (iv) only

    C:-(i), (iii) and (iv) only

    D:-Both (ii) and (v)

    Correct Answer:- Option-C

Question12:-A cost-effective principle of learning in practice is

    A:-Punishment

    B:-Re-inforcement

    C:-Rewards

    D:-Avoidance

    Correct Answer:- Option-B

Question13:-Which among the following is an intellectual disability of a learner to be addressed by a teacher ?

    A:-ADHA

    B:-Dyslexia

    C:-Autistic spectrum disorders

    D:-Mental Retardation

    Correct Answer:- Option-D

Question14:-In the five stage SOLO Taxonomy of Biggs and Collis; SOLO stands for

    A:-Structure of Observed Learning Outcomes

    B:-Strategy of Organised Learning Outcomes

    C:-Structure of Observed Learning Objectives

    D:-Sequential Organization of Learning Outcomes

    Correct Answer:- Option-A

Question15:-A web2.0 mobile app for real time continuous and comprehensive assessment of pupils engagement in class is

    A:-Go class

    B:-Edmodo

    C:-Socrative

    D:-My class talk

    Correct Answer:- Option-C

Question16:-In an item-analysis exercise, if correct to wrong response in the upper age group is 80 : 20 and that in the lower age group is 30 : 70; where N = 10 in each group; the value of D will be

    A:-2.2

    B:-2.6

    C:-2.4

    D:-2.8

    Correct Answer:- Option-B

Question17:-Action research owes to its origin to the contributions made by

    A:-Erlandson

    B:-J.W. Best

    C:-S.M.Corey

    D:-Hilda Taba

    Correct Answer:- Option-C

Question18:-The qualitative research methodology equivalent to 'reliability' in data analysis is
    A:-Internal Cohession
    B:-Predictability
    C:-Transferability
    D:-Trustworthiness
    Correct Answer:- Option-D

Question19:-Big-Data Analysis in research handles data set that are
    A:-Structured Terrabytes
    B:-Semi-structured Terrabytes
    C:-Unstructured Terrabytes
    D:-Structured Petabytes
    Correct Answer:- Option-C

Question20:-The concept of Journal Impact Factor was coined by
    A:-Robert Gunning
    B:-John Marshey
    C:-Marsha Weil
    D:-Eugene Garfield
    Correct Answer:- Option-D

Question21:-The sum of eigen values of a 2$\times$2 matrix is 5 and the product of eigen values is 6. Which of the following is the matrix ?
    A:-$[[1,1],[2,3]]$
    B:-$[[3,2],[1,1]]$
    C:-$[[3,1],[1,2]]$
    D:-$[[3,0],[0,2]]$
    Correct Answer:- Option-D

Question22:-Integrating the volume element in spherical polar coordinates gives the value
    A:-$4/3 \pi r^3$
    B:-$\pi r^2 h$
    C:-πr^2
    D:-$4\pi r^2$
    Correct Answer:- Option-A

Question23:-The Fourier transform of a Gaussian function is
    A:-Poisson function
    B:-Zero
    C:-Gaussian function
    D:-Greens function
    Correct Answer:- Option-C

Question24:-The Laplace transform of x is; (p>0)
    A:-$1/p$
    B:-$1/p^2$
    C:-$1/x^n$
    D:-$(n!)/(p^{(n+1)})$
    Correct Answer:- Option-B

Question25:-Which is the correct identity ?
    A:-$\delta_{(ii)}$ = 2
    B:-$\delta_{(ik)} \epsilon_{(ikm)}$ = 0
    C:-$\delta(x-x')$ = x
    D:-$\delta_{(xy)}$ = x^2 y
    Correct Answer:- Option-B

Question26:-The value of $\vec{L} \times \vec{L} \mu$, ($\vec{L}$ is angular momentum)
    A:-Zero
    B:-$L_{(2)}$
    C:--1

D:-

Correct Answer:- Option-D

Question27:-For a central force motion, the Lagrangian is

A:-$\frac{1}{2}m(\dot{r}^2+r^2\dot{\theta}^2)$

B:-$\frac{1}{2}m(\dot{r}^2-r^2\dot{\theta}^2)$

C:-$\frac{1}{2}m(\dot{r}^2+r^2\dot{\theta}^2)+\frac{k}{r}$

D:-$\frac{1}{2}m(r^2+\dot{r}^2\theta^2)-\frac{k}{r}$

Correct Answer:- Option-C

Question28:-For the transformation Q = aq + bp, p = cq + dp to be canonical

A:-ad + bc = 1

B:-ad – b^2 c^2 = 1

C:-ad – bc = 1

D:-ad – b^2c = 1

Correct Answer:- Option-C

Question29:-An object in the shape of a cube is moving at a speed of 0.8C. The motion is parallel to X axis. The length of the body along the Z direction is

A:-0.6 $L_{(0)}$

B:-0.36 $L_{(0)}$

C:-0.64 $L_{(0)}$

D:-$L_{(0)}$

($L_{(0)}$ is the Length at rest)

Correct Answer:- Option-D

Question30:-The Poisson bracket of three dynamical variables, [u, [v, w]] + [v, [w, u]] + [w, [u, v]] is

A:-Zero

B:---1

C:-1

D:-u

Correct Answer:- Option-A

Question31:-A curve function is written in terms of an orthonormal basis;

$|psi\rangle = \sqrt{2/6}|u_1\rangle + \sqrt{1/6}|u_2\rangle +\sqrt{3/6}|v_3\rangle$

What is the probability that the system is found to be in the state

$|v_2\rangle$ is

A:-$1/3$

B:-1

C:-$1/6$

D:-zero

Correct Answer:- Option-C

Question32:-The angular part of the wave function for 1-1-atom is written using

A:-Spherical harmonics

B:-Bessel function

C:-Hermite polynomial

D:-Newmann function

Correct Answer:- Option-A

Question33:-The value of [J+, J–] is

A:-

B:-

$$\hbar J_+$$

C:-

$$2\hbar J_+$$

D:-

$$2\hbar J_z$$

Correct Answer:- Option-D

Question34:-Which of the following is an exactly solvable time dependent problem in Quantum mechanics ?

A:-Start effect

B:-MASER

C:-Zeeman effect

D:-Harmonic Oscillator

Correct Answer:- Option-B

Question35:-Select the correct relativistic linear Hamiltonian

A:-H = `[P^2` `C^2` +`m^2_0` `C^4]``1/2`

B:-`(P^2)/(2m)`

C:-`mC^(2)` = H

D:-H = C(`alpha` .P) + `beta` `m_(0)` `C^(2)`

Correct Answer:- Option-D

Question36:-The line integral of magnetic field is

A:-Potential

B:-Current

C:-Energy

D:-Polarization

Correct Answer:- Option-B

Question37:-Magnetic vector potential is

A:-Not conservative

B:-Conservative

C:-Unipolar

D:-Endless

Correct Answer:- Option-A

Question38:-Entropy is

A:-additive quantity

B:-constant quantity

C:-conserved quantity

D:-covariant quantity

Correct Answer:- Option-A

Question39:-Bose-Einstein condensation will occur

A:-in the presence of intermolecular forces

B:-in the absence of intermolecular forces

C:-in the position space

D:-none of the above

Correct Answer:- Option-A

Question40:-Which of the following is NOT a first order phase transition ?

A:-transition from liquid to vapour

B:-transition from solid to vapour

C:-transition from ferromagnet to paramagnet

D:-melting of ice to liquid

Correct Answer:- Option-C

Question41:-To get vibrational spectra of molecule, it must involve

A:-a change in structure

B:-a change in force

C:-a change in electronic state

D:-a change in dipolemoment

Correct Answer:- Option-D

Question42:-The correct expression for rotational energy of a rotating molecule

A:-$(J(J+1)h^2)/(8\pi I)$

B:-$(J(J+1)h^2)/(8\pi^2 I)$

C:-$((J+1)h^2)/(8\pi^2 I)$

D:-$(Jh^2)/(8\pi I^2)$ `` ` `` ` `

Correct Answer:- Option-B

Question43:-Water is a

A:-spherical top molecule

B:-symmetric top molecule

C:-linear molecule

D:-asymmetric top molecule

Correct Answer:- Option-D

Question44:-How many Bravai's Lattices exists for seven crystal system ?

A:-7

B:-10

C:-13

D:-14

Correct Answer:- Option-D

Question45:-The specific heat of a superconductor

A:-decreases linearly temperature

B:-varies exponentially with temperature

C:-is independent of temperature

D:-increases linearly with temperature

Correct Answer:- Option-B

Question46:-Which is the wrong relation ?

A:-$\beta = (\alpha)/(1-\alpha)$

B:-$I_C = \beta I_B + I_{CEO}$

$$\beta = \frac{\Delta I_C}{\Delta I_B}$$

C:-

D:-$I_{CEO} = (1)/(1-\alpha)I_C$

Correct Answer:- Option-D

Question47:-For an OPAMP, the higher the value of CMRR
    A:-indicates mismatch between two input terminals
    B:-indicates low input capacitance
    C:-indicates matching between input terminals
    D:-higher offset voltage
    Correct Answer:- Option-C
Question48:-Nuclear force is
    A:-central force
    B:-spin independent force
    C:-weak force
    D:-spin dependent force
    Correct Answer:- Option-D
Question49:-The value of the boolean expression;
z = (`barA+B`) (A + B) is
    A:-B
    B:-A
    C:-`A+barA`
    D:-`barA+B`
    Correct Answer:- Option-A
Question50:-A negative Liapunov exponent measures
    A:-the rate at which a system point approaches a regular attractor
    B:-the rate at which a system point diverges from a regular attractor
    C:-the amount of sensitivity to initial conditions
    D:-the relation with poincare maps
    Correct Answer:- Option-A
Question51:-Which of the following are related to chaos ?
    A:-Breakdown of integrability
    B:-Appearance of islands
    C:-Extreme sensitivity to initial conditions
    D:-All of the above
    Correct Answer:- Option-D
Question52:-Cosmic background radiation supports
    A:-Big crunch of the universe
    B:-Steady state of the universe
    C:-Dark energy of the universe
    D:-Big Bang of the universe
    Correct Answer:- Option-D
Question53:-Which of the following is a left-handed particle ?
    A:-Antineutrino``
    B:-`pi` -measons
    C:-Neutrino
    D:-Positrons
    Correct Answer:- Option-A
Question54:-The energy gap in a semiconductor is proportional to the inverse of the square of size. The effect is a result of
    A:-Etching
    B:-Quantum confinement
    C:-Erosion
    D:-Dip pen nano-lithography
    Correct Answer:- Option-B
Question55:-Energy from gravitational field is energy obtained from
    A:-Wind
    B:-Solar cells
    C:-Tides
    D:-Coal
    Correct Answer:- Option-C
Question56:-The number of chemical shift non-equivalent protons expected in `^1`H NMR spectrum of `alpha` -plnene
    A:-7
    B:-8
    C:-9

D:-10

Correct Answer:- Option-D

Question57:-Example of a boson

    A:-Photon

    B:-Electron

    C:-Neutron

    D:-Positron

    Correct Answer:- Option-A

Question58:-The molecule that has $S_{(6)}$ symmetry element is

    A:-$B_{(2)}H_{(6)}$

    B:-$CH_{(4)}$

    C:-$PH_{(3)}$

    D:-$SF_{(6)}$

    Correct Answer:- Option-D

Question59:-Calculate the dipole moment of NaCl in the gas phase assuming equal and opposite charges equal to the proton charges. The equilibrium internuclear distance is 236 pm.

    A:-38×10^{-20} Cm

    B:-28×10^{-15} Cm

    C:-38×10^{-30} Cm

    D:-18×10^{-30} Cm

    Correct Answer:- Option-C

Question60:-In the iodometric titration of sodium thiosulfate ($Na_{(2)}S_{(2)}O_{(3)}$) with acidic dichromate solution, 25 mL of 0.1 M dichromate requires 25 mL of 'x' M thiosulfate. The value of 'x' is

    A:-0.2

    B:-0.1

    C:-0.6

    D:-0.4

    Correct Answer:- Option-D

Question61:-The loss of an alkene fragment by a cyclic rearrangement of acarbonyl compound with Γ-hydrogen is termed as

    A:-Beckman rearrangement

    B:-Lossen rearrangement

    C:-Cope rearrangement

    D:-Mc Lafferty rearrangement

    Correct Answer:- Option-D

Question62:-How many peaks do you expect in the proton decoupled CMR spectra of cycloheptane and t-butyl alcohol ?

    A:-1, 2

    B:-2, 4

    C:-1, 3

    D:-2, 2

    Correct Answer:- Option-A

Question63:-The heterocyclic ring present in the amino acid histidine is

    A:-Pyridine

    B:-Tetrahydrofuran

    C:-Indole

    D:-Imidazole

    Correct Answer:- Option-D

Question64:-The co-enzyme that is involved in the reduction of a double bond in fatty acid bisyntheis is

    A:-NADH

    B:-Biotin

    C:-Pyridoxal

    D:-FADH2

    Correct Answer:- Option-D

Question65:-Ground state term symbols for Lanthanum and Uranium atom are

    A:-$^2L_{(3/2)}, ^5L_{(6)}$

    B:-$^2D_{(2/3)}, ^6L_{(5)}$

    C:-$^2D_{(3/2)}, ^5L_{(6)}$

    D:-$^1D_{(3/2)}, ^5L_{(5)}$

    Correct Answer:- Option-C

Question66:-The simultaneous Eigen functions of angular momentum operators $L^{(2)}$ and $L_{(z)}$ are

A:-all of 2_s, $2p_x$, $2p_y$ and $2p_z$ orbitals

B:-only 2_s, $2p_x$, $2p_y$ orbitals

C:-only 2_s and $2p_z$ orbitals

D:-only $2p_z$ orbital

Correct Answer:- Option-C

Question67:-The point group symmetry of $CH_2 = C = CH_2$ is

A:-C_{2h}

B:-C_{2v}

C:-D_{2h}

D:-D_{2d}

Correct Answer:- Option-D

Question68:-The molecule that has S_6 symmetry element is

A:-B_2H_6

B:-CH_4

C:-PH_3

D:-SF_6

Correct Answer:-Question Cancelled

Question69:-An element with density 11.2 g/cm^3 forms an fcc lattice with edge length of 4×10^{-8} cm. Calculate the atomic mass of the element.

A:-120.8

B:-240.4

C:-98

D:-107.9

Correct Answer:- Option-D

Question70:-The crystal with metal deficiency defect is

A:-NaCl

B:-ZnO

C:-FeO

D:-KCl

Correct Answer:- Option-C

Question71:-Which one of the following is used as piezo electric material ?

A:-Silica gel

B:-Graphite

C:-Kieselghur

D:-Quartz

Correct Answer:- Option-D

Question72:-Standard electrode potentials of three metals X, Y, Z are –1.2V, +.5V and –3V respectively. The reducing power of these metals will be

A:-Y>Z>

B:-Y>X>Z

C:-Z>X>Y

D:-X>Y>Z

Correct Answer:- Option-C

Question73:-The most populated rotational state for HCl (B = 8.5 cm^{-1}) at 300 K is

A:-4

B:-3

C:-1

D:-8

Correct Answer:- Option-B

Question74:-The value of $d_{(111)}$ in a cubic crystal is 325.6 pm. The value of $d_{(333)}$ is

A:-208 pm

B:-308 pm

C:-108.5 pm

D:-420 pm

Correct Answer:- Option-C

Question75:-The molecule active in rotational microwave, infrared absorption as well as rotational Raman spectra is

A:-CO_2

B:-SF_6

C:-HCl

D:-`N_(2)`

Correct Answer:- Option-C

Question76:-Which of the following electrolyte have maximum Flocculation Value ?

A:-`K_(2)SO_(4)`

B:-`(NH_(4))_3PO_(4)`

C:-`Na_(2)` S

D:-NaCl

Correct Answer:- Option-D

Question77:-The equilibrium constant of the following redox reactions at 298 K is `1xx10^(8)` `2Fe^(3+)` (aq) + 2I-(aq)=== `2fe^(2)` +(aq) + `I_(2)` (s)

If the standard reduction potential of iodine becoming iodine is +.54V, what is the standard reduction potential of `Fe^(3+)`/`Fe^(2+)` ?

A:--1.006V

B:-+.77V

C:--.77V

D:--.652V

Correct Answer:- Option-B

Question78:-A first order reaction is 50% complete in 69.3 minutes. Time required for 90% completion for this reaction is

A:-100 minutes

B:-230 minutes

C:-2303 minutes

D:-125 minutes

Correct Answer:- Option-B

Question79:-An organic compound with the molecular formula `C_(9)H_(10)`O forms 2, 4-DNP derivative, reduces Tollen's reagent and undergoes Cannizzaro reaction. On vigorous oxidation, it gives 1, 2-bezendicarboxylic acid. Identify the compound.

A:-2-methyl Benzaldehyde

B:-3-ethyl Benzaldehyde

C:-2-ethyl Benzaldehyde

D:-4-ethyl Benzaldehyde

Correct Answer:- Option-C

Question80:-The reaction that involves the formation of both C-C and C-O bonds is

A:-Diels-Alder reaction

B:-Darzen's glycidic ester condensation

C:-Aldol reaction

D:-Beckmann rearrangement

Correct Answer:- Option-B

Question81:-Amongst the following which is not isolobal pairs ?

A:-`Mn(CO)_(5)`, `CH_(3)`

B:-`Fe(CO)_(4)`, O``

C:-`Co(CO)_(3)`, `R_(2)Si`

D:-`Mn(CO)_(5)`, RS``

Correct Answer:- Option-C

Question82:-The ligand system present in Vitamin `B_(12)` is

A:-Porphyrin

B:-Corrin

C:-Phthalocyanine

D:-Crown ether

Correct Answer:- Option-B

Question83:-Match the metalloproteins in Column A with its biological and metal center in Column B.

Column A	Column B
(a) Hemoglobin	(i) Electron carrier and iron
(b) Cytochrome b	(ii) Electron carrier and copper
(c) Vitamin `B_(12)`	(iii) `O_(2)` transport and copper
(d) Hemocyanin	(iv) Group transfer reactions and cobalt
	(v) `O_(2)` storage and cobalt
	(vi) `O_(2)` transport and iron

The correct match is

A:-(a)-(vi); (b)-(i); (c)-(iv) and (d)-(iii)

B:-(a)-(v); (b)-(i); (c)-(iv) and (d)-(iii)

C:-(a)-(iv); (b)-(v); (c)-(i) and (d)-(ii)

D:-(a)-(v); (b)-(vi); (c)-(ii) and (d)-(iv)

Correct Answer:- Option-A

Question84:-The light pink color of $[Co(H_2O)_6]^{2+}$ and the deep blue color of $[CoCl_4]^{2-}$ are due to

A:-MLCT transition in the first and d-d transition in the second

B:-LMCT transition in both

C:-d-d transitions in both

D:-d-d transition in the first and MLCT transition in the second

Correct Answer:- Option-C

Question85:-Patients suffering from Wilson's disease have

A:-Low level of Cu-Zn superoxide dismutase

B:-High level of Cu-Zn superoxide dismutase

C:-Low level of copper-storage protein, ceruloplasmin

D:-High level of copper-storage protein, ceruloplasmin

Correct Answer:- Option-C

Question86:-The carbon-14 activity of an old wood sample is found to be 14.2 disintegrations $min^{-1}g^{-1}$. Calculate age of oldwood sample, if for a fresh wood sample carbon-14 activity is 15.3 disintegrations $min^{-1}g^{-1}(t_{1/2}$ carbon–14 is 5730 years), is

A:-5,000 years

B:-4,000 years

C:-877 years

D:-617 years

Correct Answer:- Option-D

Question87:-In a polarographic estimation, the limiting currents (μA) were 0.15, 4.65, 9.15 and 27.15 when concentration (mM) of Pb(II) were 0, 0.5, 1.0 and 3.0 respectively. An unknown solution of Pb(II) gives a limiting current of 13.65 μA. Concentration of Pb(II) in the unknown is

A:-1.625 mM

B:-1.208 mM

C:-1.768 mM

D:-1.500 mM

Correct Answer:- Option-D

Question88:-The number of valence electrons provided $[Ru(CO)_3]$ fragment towards cluster bonding is

A:-1

B:-12

C:-8

D:-2

Correct Answer:- Option-D

Question89:-0.106 mg of stearic acid (M = 284 g mol^{-1}) is found to cover 500 cm^2 of water surface at the point where surface pressure just begins to rise sharply. Estimate the cross sectional area a, per stearic acid molecule and thickness t, of the surface film of stearic acid on water. Density of stearic acid = 0.85 g cm^{-3}.

A:-3.5 nm

B:-2.5 nm

C:-1.9 nm

D:-.05 cm

Correct Answer:- Option-B

Question90:-For Daniel cell calculate ΔG, if E = 1.10v at 250 C and D = 2, F = 96500 C

A:-185.4 KJ

B:-200.2 KJ

C:-212.3 KJ

D:-256.2 KJ

Correct Answer:- Option-C

Question91:-Which of the following expression does not figure in the Preamble of Indian Constitution ?

A:-Republic

B:-Socialist

C:-Federal

D:-Secular

Correct Answer:- Option-C

Question92:-Which was described by Dr. B.R. Ambedkar as the 'Heart and Soul of the Constitution' ?

A:-Right to Equality

B:-Right to Constitutional Remedies

C:-Right to Freedom

D:-Right to Freedom of Religion

Correct Answer:- Option-B

Question93:-The Article of Indian Constitution which deals with Uniform Civil Code ?

A:-44

B:-48A

C:-43A

D:-47

Correct Answer:- Option-A

Question94:-The total number of ministers including the Prime Minister in the Council of Ministers should not exceed _____ percent of the total membership of the Lok Sabha.

A:-7

B:-10

C:-20

D:-15

Correct Answer:- Option-D

Question95:-Which Article of Indian Constitution deals with Amendment ?

A:-368

B:-352

C:-358

D:-315

Correct Answer:- Option-A

Question96:-Which of the following is a small deposit scheme for girl child ?

A:-Pradhan Mantri Suraksha Bima Yojana

B:-Pradhan Mantri Jeevan Jyothi Bima

C:-Pradhan Mantri Sukanya Samridhi Yojana

D:-Pradhan Mantri Fasal Bima Yojana

Correct Answer:- Option-C

Question97:-World Blood Donor Day is celebreated each year throughout the world on

A:-17th August

B:-14th June

C:-9th September

D:-20th February

Correct Answer:- Option-B

Question98:-National Rural Employment Guarantee Scheme was extended to the State of Jammu and Kashmir in the year

A:-2007

B:-2005

C:-2006

D:-2008

Correct Answer:- Option-A

Question99:-National Food Security Act came into force on

A:-23rd July, 2013

B:-5th July, 2013

C:-15th August, 2013

D:-1st June, 2013

Correct Answer:- Option-B

Question100:-The definition of 'Child' in the 'Protection of Women from Domestic Violence Act 2005' is

A:-Any person below the age of 15 years

B:-Any person below the age of 14 years

C:-Any person below the age of 7 years

D:-Any person below the age of 18 years

Correct Answer:- Option-D